

# Tributyltin hydride-mediated tandem reactions of dehydroalanines leading to $\alpha$ -substituted pyrroglutamates

S. Richard Baker,<sup>a</sup> Karen Goodall,<sup>b</sup> Andrew F. Parsons<sup>b\*</sup> and Michelle Wilson<sup>b</sup>

<sup>a</sup>Eli Lilly and Company Ltd, Lilly Research Centre, Erl Wood Manor, Windlesham, Surrey, GU20 6PH, UK

<sup>b</sup>Department of Chemistry, University of York, Heslington, York, YO10 5DD, UK

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Tributyltin hydride-mediated cyclisation of dehydroalanines produces an intermediate captodative radical, which can be trapped by reaction with oxygen- or carbon-centred radicals or (principally) electron-poor alkenes, to provide a quick approach to a variety of  $\alpha$ -substituted pyrroglutamates.

One important method for preparing substituted pyrrolidinones of biological importance, centres on the radical cyclisation of unsaturated haloamides.<sup>1</sup> These reactions generally involve cyclisation of a carbamoylmethyl radical on to an *N*-allyl double bond, in a (favoured) 5-*exo-trig* pathway, and this has been shown to provide a mild and flexible approach to a variety of pyrrolidinones.<sup>2</sup> More recently, however, an alternative approach to pyrrolidinones has been developed, which centres on the 5-*endo-trig* radical cyclisation of halo-enamides bearing an *N*-vinyl double bond.<sup>3</sup> This cyclisation is unusual (disfavoured) in that the initial carbamoylmethyl radical reacts to form a 5- rather than a 4-membered (or  $\beta$ -lactam) ring. The 4-membered ring would normally be expected from a (favoured) 4-*exo-trig* cyclisation. We have recently demonstrated that this method of cyclisation can be used to prepare pyrroglutamates **4** from halo-dehydroalanines of type **1** (Scheme 1).<sup>4</sup> The regioselective formation of a pyrroglutamate ring could result from a reversible cyclisation, which leads to the predominant formation of captodative radical **3**.

As the radical cyclisation of **1** proceeds *via* captodative radical **3**, this offers a potential method for preparing  $\alpha$ -substituted pyrroglutamates in one-pot or tandem reactions. Following our studies on the intramolecular reaction of radicals of type **3**,<sup>6</sup> this paper reports the (entropically more difficult) intermolecular trapping of these types of radical to form conformationally restricted  $\alpha$ -substituted pyrroglutamates.<sup>7</sup> Radicals of type **3** are shown to react with carbon- or oxygen-

centred radicals to yield  $\alpha$ -substituted pyrroglutamates, such as **7** and **25**, in 13 and 32% yield, respectively. Reactions with chiefly electron-poor alkenes, such as methyl methacrylate, allow the direct formation of C-2 alkylated pyrroglutamates such as **16**. During the course of this work, an unusual intermolecular radical addition followed by intramolecular cyclisation sequence has been shown to provide a quick approach to oxo-azepane derivatives such as **18**. This involves regioselective addition of carbamoylmethyl radical **2** to methyl methacrylate, to form tertiary radical **20**, before a 5-*endo* cyclisation can take place (Scheme 5). Radical **20** then undergoes a 7-*endo-trig* cyclisation to produce captodative radical **21**, which leads to the formation of oxo-azepane **18** on hydrogen-atom abstraction from tributyltin hydride.

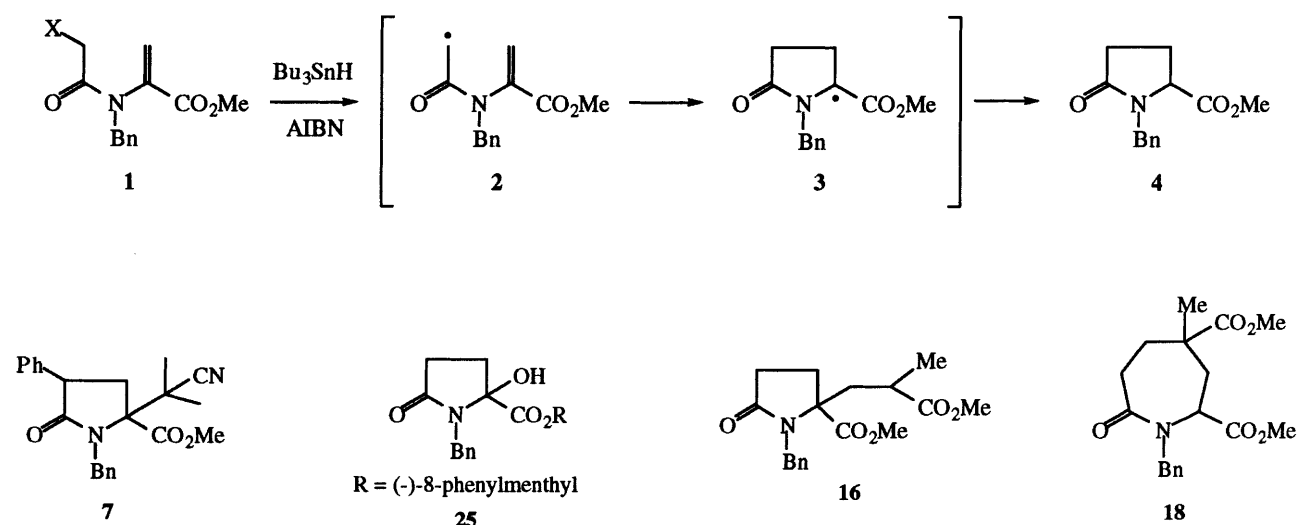
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Techniques used: TLC, FTIR, <sup>1</sup>H and <sup>13</sup>C NMR, LRMS, HRMS

References: 12

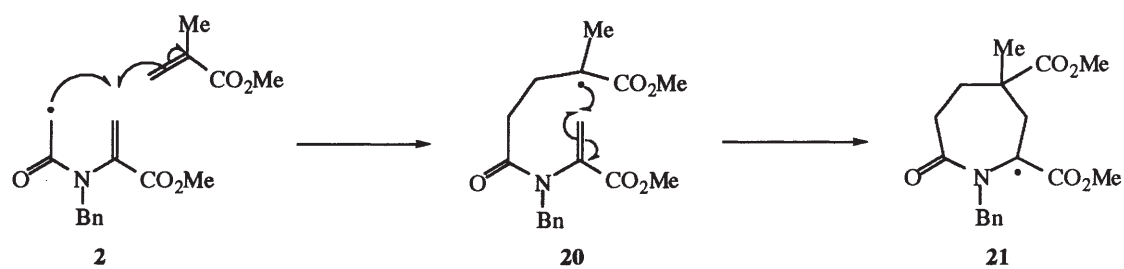
Tables: 2

Schemes: 7



Scheme 1

\* To receive any correspondence. E-mail: afp2@york.ac.uk



Scheme 5

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#### References cited in this synopsis

- 1 (a) B. Giese, B. Kopping, T. Göbel, J. Dickhaut, G. Thoma, K.J. Kulicke and F. Trach, *Organic Reactions*, 1996, **48**, 301. (b) F. Aldabbagh and W.R. Bowman, *Contemp. Org. Synth.*, 1997, 261. (c) W.R. Bowman, C.F. Bridge and P. Brookes, *J. Chem. Soc., Perkin Trans. 1*, 2000, 1.
- 2 For recent examples see: (a) J.S. Bryans, J.M. Large and A.F. Parsons, *J. Chem. Soc., Perkin Trans. 1*, 1999, 2897. (b) J.S. Bryans, J.M. Large and A.F. Parsons, *J. Chem. Soc., Perkin Trans. 1*, 1999, 2905.
- 3 For recent examples see: (a) H. Ishibashi, M. Higuchi, M. Ohba and M. Ikeda, *Tetrahedron Lett.*, 1998, **39**, 75. (b) M. Ikeda, S. Ohtani, T. Yamamoto, T. Sato and H. Ishibashi, *J. Chem. Soc., Perkin Trans. 1*, 1998, 1763.
- 4 (a) K. Goodall and A.F. Parsons, *J. Chem. Soc., Perkin Trans. 1*, 1994, 3257. (b) K. Goodall and A.F. Parsons, *Tetrahedron*, 1996, **52**, 6739.
- 6 S.R. Baker, K.I. Burton, A.F. Parsons, J-F. Pons and M. Wilson, *J. Chem. Soc., Perkin Trans. 1*, 1999, 427.
- 7 Part of this work has been reported previously: S.R. Baker, A.F. Parsons and M. Wilson, *Tetrahedron Lett.*, 1998, **39**, 2815.